

## QUALITATIVE ANALYSIS OF SIMPLE SALTS TABLES%0A

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### **Qualitative Analysis Identifying Anions and Cations**

Qualitative analysis is used to identify and separate cations and anions in a sample substance. Unlike quantitative analysis, which seeks to determine the quantity or amount of sample, qualitative analysis is a descriptive form of analysis. In an educational setting, the concentrations of the ions to be identified are approximately 0.01 M in an aqueous solution. The 'semimicro' level of qualitative

analysis employs methods used to detect 1-2 mg of an ion in 5 mL of solution.

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### **Qualitative Analysis Test for Cations**

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### **NIT 7 YSTEMATIC QUALITATIVE NALYSIS**

Qualitative analysis of inorganic salts means the identification of cations and anions present in the salt or a mixture of salts. Inorganic salts may be obtained by complete or partial neutralisation of acid with base or vice-versa. In the formation of a salt, the part contributed by the acid is called anion and the part contributed by the base is called cation. For example, in the salts CuSO

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### **Qualitative Analysis Wired Chemist**

There are two general situations in which qualitative analysis is used - in the identification of a simple salt, or the identification of multiple cations in a solution. Identifying a Simple Salt. The basic testing procedure for identifying a salt is as follows. Appearance of compound. The compound will most likely be in solid form. Note the color and shape of the crystals. Ionic compounds

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### **Skills Practice Lab MICROSCALE Simple Qualitative Analysis**

There are hundreds of simple qualita-tive tests such as these to distinguish among a few possibilities. Qualitative tests are important to the forensic chemist, one who is interested in solving crimes, but they have been largely replaced by instrumental analysis, which is fast and requires a very small sample. For example, instrumental analy-sis can detect as little as 1 10 11 g of 9

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